

# APPROACHES TO THE SYNTHESIS OF MANZAMINE A. SYNTHESIS OF THE $\beta$ -CARBOLINE-BEARING ABCE RING SYSTEM

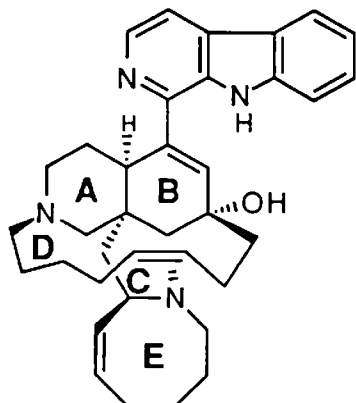
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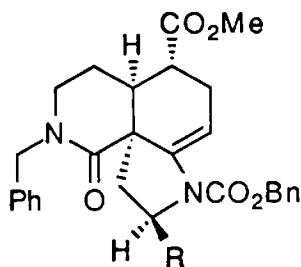
**Abstract :** The 8-membered ring E has been introduced onto a strategically functionalized chiral pyrrolo[2,3-*l*]isoquinoline derivative by a combination of a Wittig coupling and amide cyclization. The resulting tetracyclic structure has been converted to the ABCE- $\beta$ -carboline ring system of manzamine A.

The novel structure and significant biological activity of manzamine A **1** - an alkaloid isolated from sea sponges found in the Okinawan waters (1) - has drawn keen attention in connection with its biosynthetic origin (2) and chemical synthesis (3).

In a retrosynthetic analysis of the alkaloid, we, and a number of other groups, have recognized that the ABC ring system of manzamine A constitutes a core structure upon which the remaining rings of the target compound may be elaborated. In this context, we reported the first synthesis of the strategically functionalized chiral pyrrolo[2,3-*l*]isoquinoline intermediate **2** (3d). We now present the elaboration of **2** to the ABCE- $\beta$ -carboline ring system of manzamine A.

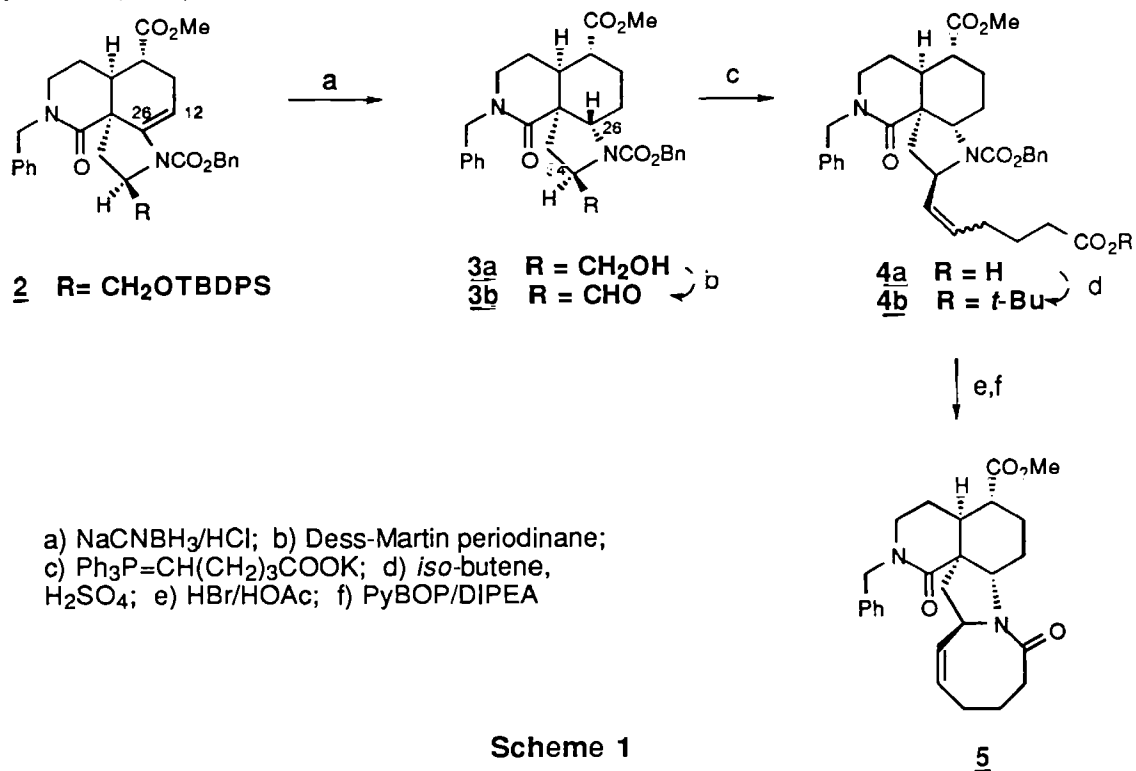


Manzamine A **1**



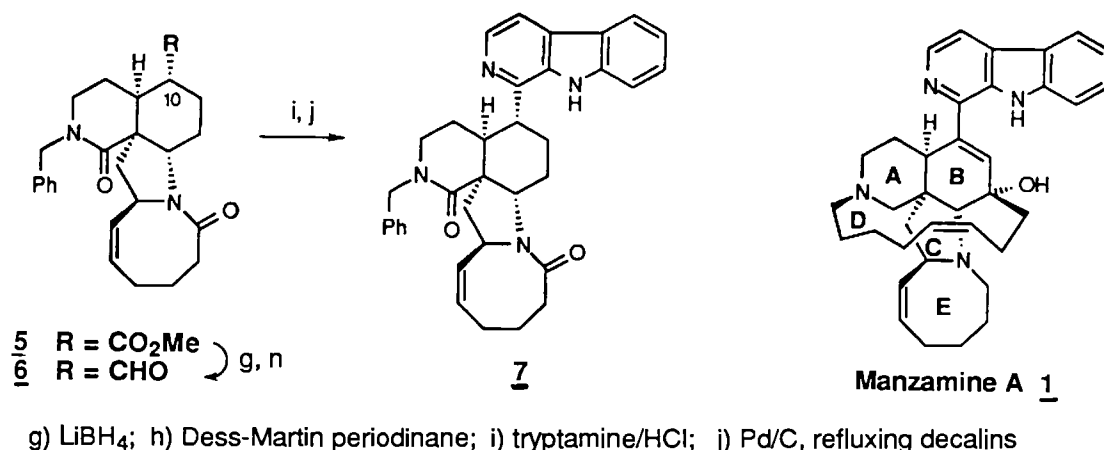
**2** R= CH<sub>2</sub>OTBDPS

In the present study, the C-(12)-C-(26) double bond of intermediate **2** [manzamine numbering (1b)], was reduced in order to simplify the functional character of the system. During the reduction (NaCNBH<sub>3</sub>/HCl), the *t*-butyldiphenylsilyl (TBDPS) group of **2** was cleaved to give alcohol **3a** (80%) (4) (**Scheme 1**). Oxidation of **3a** using the Dess-Martin periodinane (5) provided the corresponding aldehyde **3b**, in good yield (94%). The  $\beta$ -configuration of the C(26)-hydrogen, in **3b**, was derived from an NOE experiment (6). A Wittig coupling of **3b** with Ph<sub>3</sub>P=CH(CH<sub>2</sub>)<sub>3</sub>COOK, under conditions suited to selectively give the *Z*-olefin, resulted in an isomeric mixture defined by **4a** (46%), containing the *Z*-isomer as the major product (*Z/E*, 4/1). The *E/Z* ratio in the mixture of **4a** was determined by the <sup>1</sup>H NMR spectrum of the corresponding *t*-butyl esters **4b**. The low yield of the Wittig condensation is attributed to steric hindrance by the *N*-benzyl substituent. This feature is convincingly evidenced in the molecular model of **4b** and is further attested by the fact that a similar Wittig reaction of both proline carboxaldehyde (7) and an intermediate analogous to **4b**, but lacking the *N*-benzyl group (3o), proceeds in good yield. The isomeric mixture of **4a** was subjected to removal of the benzyloxycarbonyl protecting group (HBr/HOAc); treatment with the PyBOP peptide coupling reagent (8) gave the homochiral ABCE ring system **5** (45%).



Scheme 1

The introduction of the  $\beta$ -carboline moiety on the tetracycle **5** was achieved using a standard Pictet-Spengler sequence (**Scheme 2**) (9). To this end, the ester group at C-10 was reduced and subsequently oxidized to afford aldehyde **6**. Condensation of **6** with tryptamine, followed by cyclization using HCl in benzene and a final oxidative aromatization step (Pd/C) yielded the  $\beta$ -carboline derivative **7**, the structure of which was attested by its spectral data (10). The overall yield for this unoptimized five step sequence was 15%.



Scheme 2

We have recently developed an efficient strategy for the elaboration of ring D on the tricyclic intermediate **2** (R=H) in our laboratory (3r). The application of the methodologies for constructing rings D, E and the  $\beta$ -carboline nucleus, onto the chiral pyrrolo[2,3-]isoquinoline core, is currently in progress.

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- (10) Selected data for **7**:  $[\alpha]_D = +8.9$  ( $c = 0.2$ ,  $\text{CHCl}_3$ ); MS (FAB) 545 ( $\text{M}^+ + \text{H}$ , 73%), 448 (61), 279 (80), 250 (41), 91 (100);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ) 1.30 (m, 2H), 1.64-1.83 (m, 3H), 1.88-2.04 (m, 4H), 2.27-2.40 (m, 2H), 2.55 (dd, 1H,  $J = 7.1, 12.2$  Hz), 2.75 (m, 2H), 3.00-3.40 (m, 3H), 3.52 (m, 1H), 3.58 (d, 1H,  $J = 13.9$  Hz), 3.67 (m, 1H), 4.86 (m, 1H), 5.50 (m, 1H), 5.65 (m, 1H), 5.73 (d, 1H,  $J = 13.9$  Hz), 6.92 (br s, 1H), 7.23-7.30 (m, 2H), 7.34 (d, 1H,  $J = 8.3$  Hz), 7.40-7.70 (m, 5H), 7.74 (d, 1H,  $J = 5.3$  Hz), 8.03 (d, 1H,  $J = 7.8$  Hz), 8.36 (d, 1H,  $J = 5.3$  Hz)

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