

# APPROACHES TO THE SYNTHESIS OF MANZAMINE A. SYNTHESIS OF THE $\beta$ -CARBOLINE-BEARING ABCE RING SYSTEM

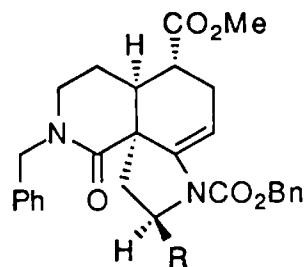
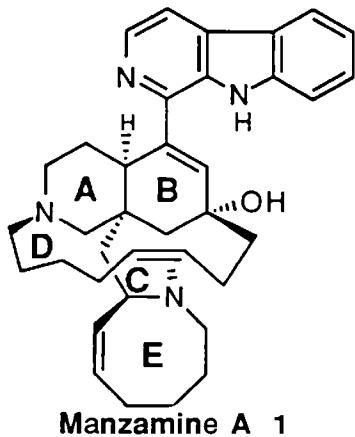
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**Abstract :** The 8-membered ring E has been introduced onto a strategically functionalized chiral pyrrolo[2,3-*i*]isoquinoline derivative by a combination of a Wittig coupling and amide cyclization. The resulting tetracyclic structure has been converted to the ABCE- $\beta$ -carboline ring system of manzamine A.

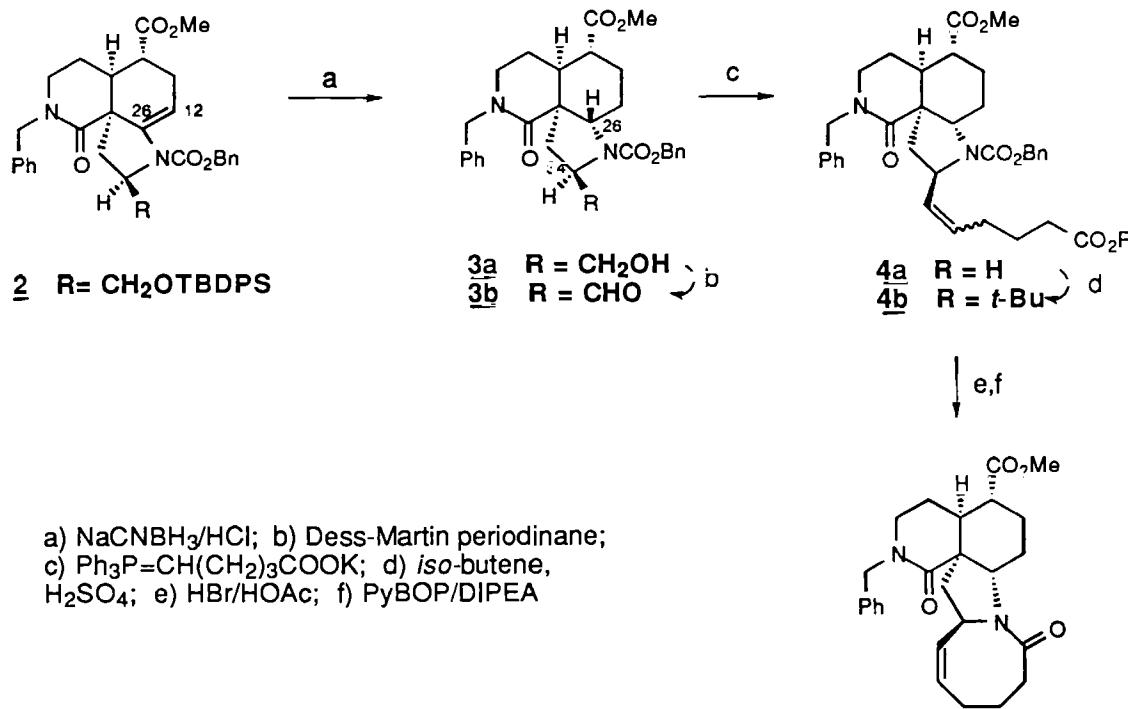
The novel structure and significant biological activity of manzamine A 1 - an alkaloid isolated from sea sponges found in the Okinawan waters (1) - has drawn keen attention in connection with its biosynthetic origin (2) and chemical synthesis (3).

In a retrosynthetic analysis of the alkaloid, we, and a number of other groups, have recognized that the ABC ring system of manzamine A constitutes a core structure upon which the remaining rings of the target compound may be elaborated. In this context, we reported the first synthesis of the strategically functionalized chiral pyrrolo[2,3-*i*]isoquinoline intermediate 2 (3d). We now present the elaboration of 2 to the ABCE- $\beta$ -carboline ring system of manzamine A.



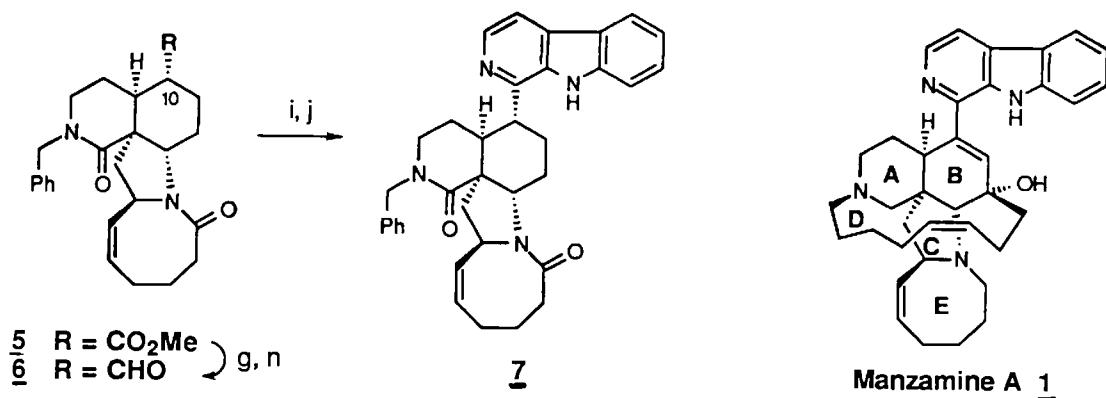
2 R= CH<sub>2</sub>OTBDPS

In the present study, the C-(12)-C-(26) double bond of intermediate 2 [manzamine numbering (1b)], was reduced in order to simplify the functional character of the system. During the reduction (NaCNBH<sub>3</sub>/HCl), the *t*-butyldiphenylsilyl (TBDPS) group of 2 was cleaved to give alcohol 3a (80%) (4) (Scheme 1). Oxidation of 3a using the Dess-Martin periodinane (5) provided the corresponding aldehyde 3b, in good yield (94%). The  $\beta$ -configuration of the C(26)-hydrogen, in 3b, was derived from an NOE experiment (6). A Wittig coupling of 3b with Ph<sub>3</sub>P=CH(CH<sub>2</sub>)<sub>3</sub>COOK, under conditions suited to selectively give the *Z*-olefin, resulted in an isomeric mixture defined by 4a (46%), containing the *Z*-isomer as the major product (*Z/E*, 4/1). The *E/Z* ratio in the mixture of 4a was determined by the <sup>1</sup>H NMR spectrum of the corresponding *t*-butyl esters 4b. The low yield of the Wittig condensation is attributed to steric hindrance by the *N*-benzyl substituent. This feature is convincingly evidenced in the molecular model of 4b and is further attested by the fact that a similar Wittig reaction of both proline carboxaldehyde (7) and an intermediate analogous to 4b, but lacking the *N*-benzyl group (3o), proceeds in good yield. The isomeric mixture of 4a was subjected to removal of the benzyloxycarbonyl protecting group (HBr/HOAc); treatment with the PyBOP peptide coupling reagent (8) gave the homochiral ABCE ring system 5 (45%).



Scheme 1

The introduction of the  $\beta$ -carboline moiety on the tetracycle 5 was achieved using a standard Pictet-Spengler sequence (Scheme 2) (9). To this end, the ester group at C-10 was reduced and subsequently oxidized to afford aldehyde 6. Condensation of 6 with tryptamine, followed by cyclization using HCl in benzene and a final oxidative aromatization step (Pd/C) yielded the  $\beta$ -carboline derivative 7, the structure of which was attested by its spectral data (10). The overall yield for this unoptimized five step sequence was 15%.



Scheme 2

We have recently developed an efficient strategy for the elaboration of ring D on the tricyclic intermediate **2** (R=H) in our laboratory (3r). The application of the methodologies for constructing rings D, E and the  $\beta$ -carboline nucleus, onto the chiral pyrrolo[2,3-*i*]isoquinoline core, is currently in progress.

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(4) Satisfactory  $^1\text{H}$  NMR, IR, and mass spectral data were obtained for all new compounds using chromatographically homogeneous samples

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(6) The  $\beta$ -configuration of C(26)-H in **3b** is attested by the absence of an enhancement of the C(26)-H signal upon irradiation of C(34)-H ( $\delta$  4.06 ppm, benzene-d<sub>6</sub>, 343 K) in an NOE experiment. Based on molecular models, an  $\alpha$ -configuration of C(26)-H would have resulted in a positive effect in the NOE experiment

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(10) Selected data for **7**:  $[\alpha]_D = +8.9$  ( $c = 0.2$ ,  $\text{CHCl}_3$ ); MS (FAB) 545 ( $\text{M}^++\text{H}$ , 73%), 448 (61), 279 (80), 250 (41), 91 (100);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ) 1.30 (m, 2H), 1.64-1.83 (m, 3H), 1.88-2.04 (m, 4H), 2.27-2.40 (m, 2H), 2.55 (dd, 1H,  $J=7.1, 12.2$  Hz), 2.75 (m, 2H), 3.00-3.40 (m, 3H), 3.52 (m, 1H), 3.58 (d, 1H,  $J=13.9$  Hz), 3.67 (m, 1H), 4.86 (m, 1H), 5.50 (m, 1H), 5.65 (m, 1H), 5.73 (d, 1H,  $J=13.9$  Hz), 6.92 (br s, 1H), 7.23-7.30 (m, 2H), 7.34 (d, 1H,  $J=8.3$  Hz), 7.40-7.70 (m, 5H), 7.74 (d, 1H,  $J=5.3$  Hz), 8.03 (d, 1H,  $J=7.8$  Hz), 8.36 (d, 1H,  $J=5.3$  Hz)

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